

FORMATIVE ASSESSMENTS THAT SOLIDIFY STUDENT LEARNING

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Learning Goals

By the end of this session, you will:

be able to **describe and implement a formative assessment strategy**

be able to **describe the benefits of this strategy** from the students and instructor's perspectives

know where to find resources to help implement this assessment strategy

Why do we assess?

Rationales for assessing

Opportunity for
instructor to evaluate
student learning
(84%)

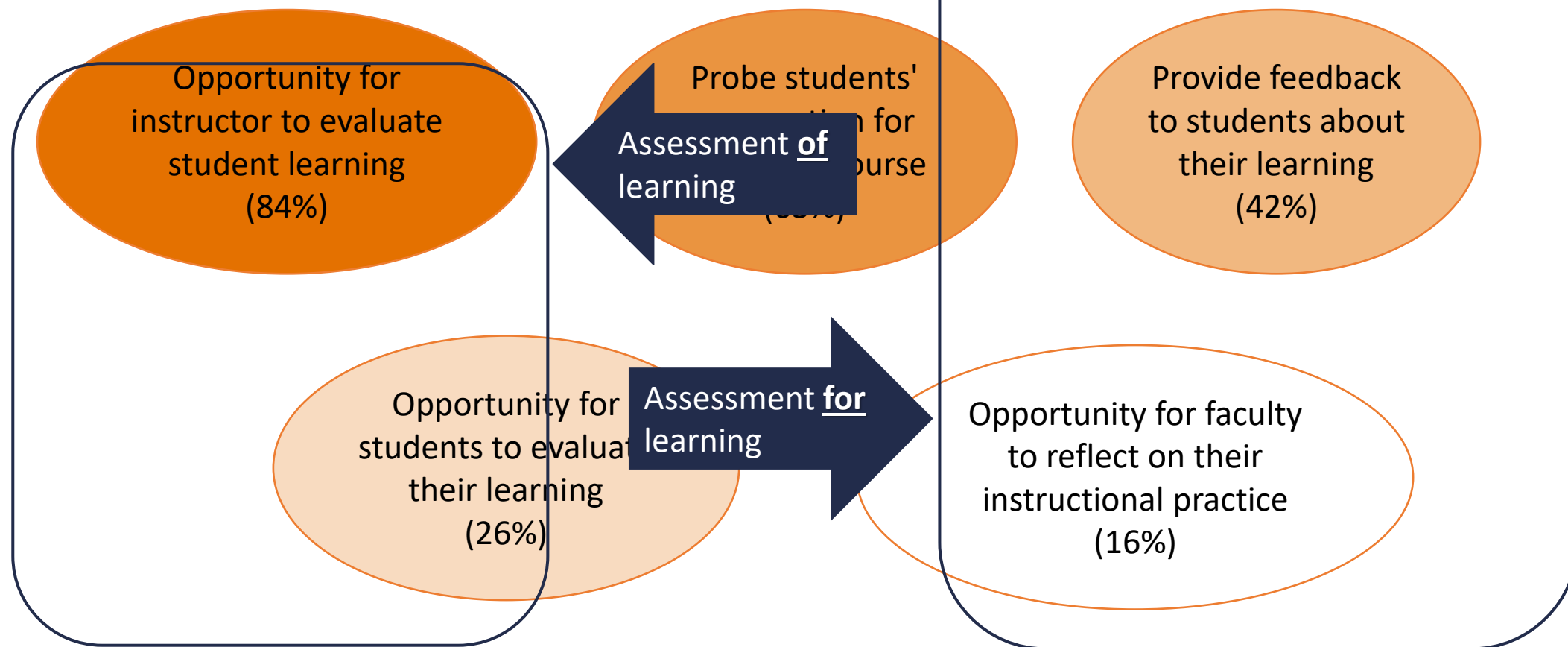
Probe students'
preparation for
their next course
(63%)

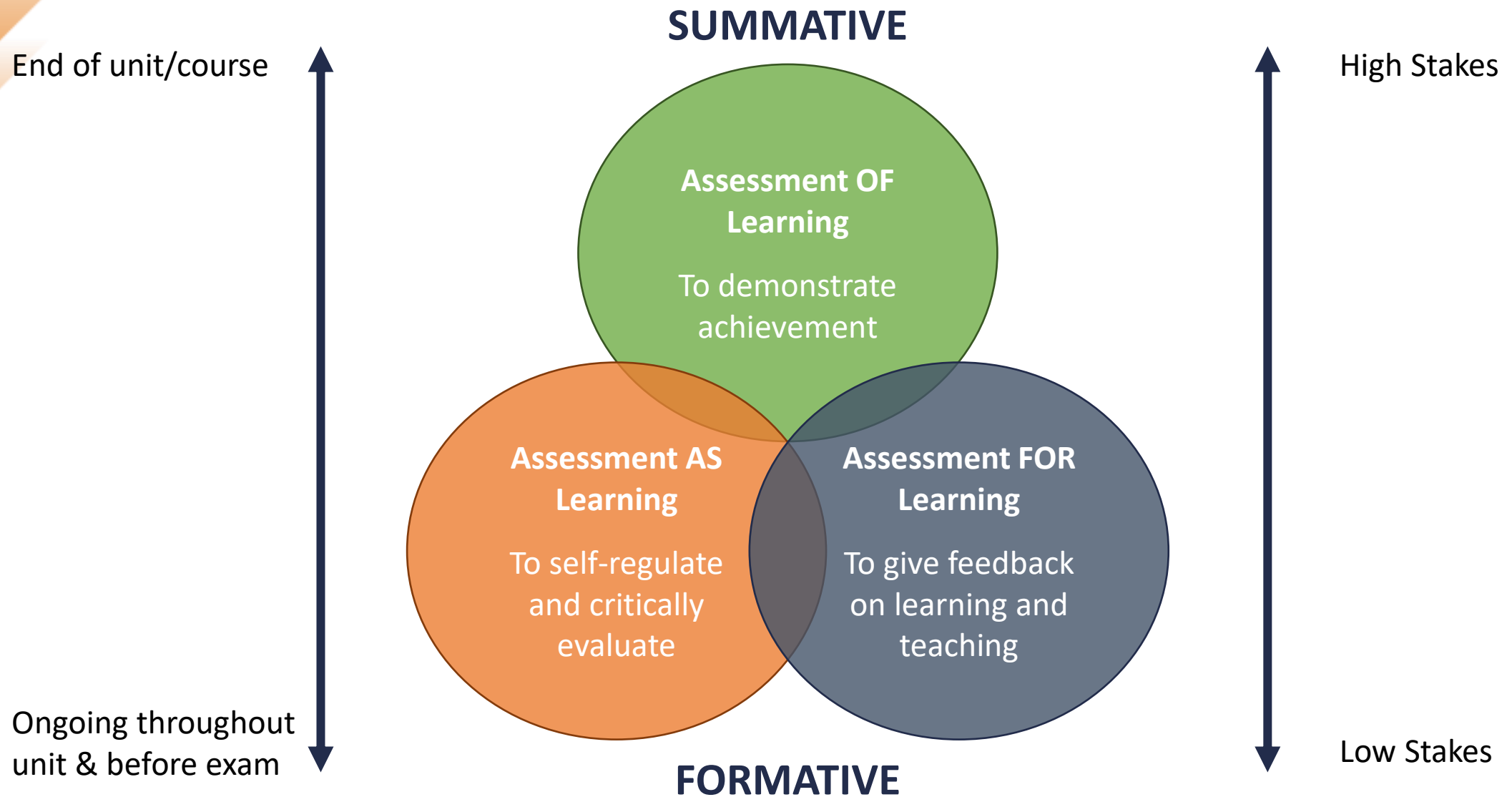
Provide feedback
to students about
their learning
(42%)

Opportunity for
students to evaluate
their learning
(26%)

Opportunity for faculty
to reflect on their
instructional practice
(16%)

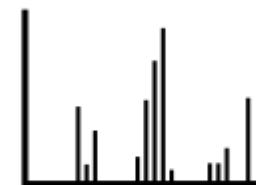
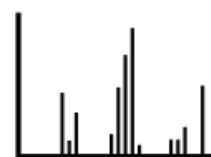
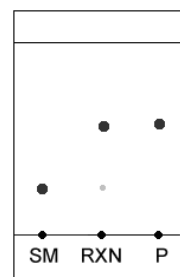
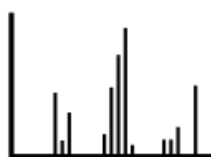
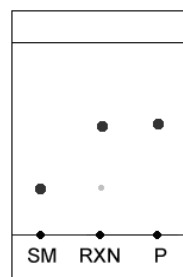
Rationales for assessing





Assessment FOR Learning aligns with our practices in the lab!

Compound A $\xrightarrow{\text{Reagents}}$ Compound B $\xrightarrow{\text{Reagents}}$ Compound C $\xrightarrow{\text{Reagents}}$ Compound D

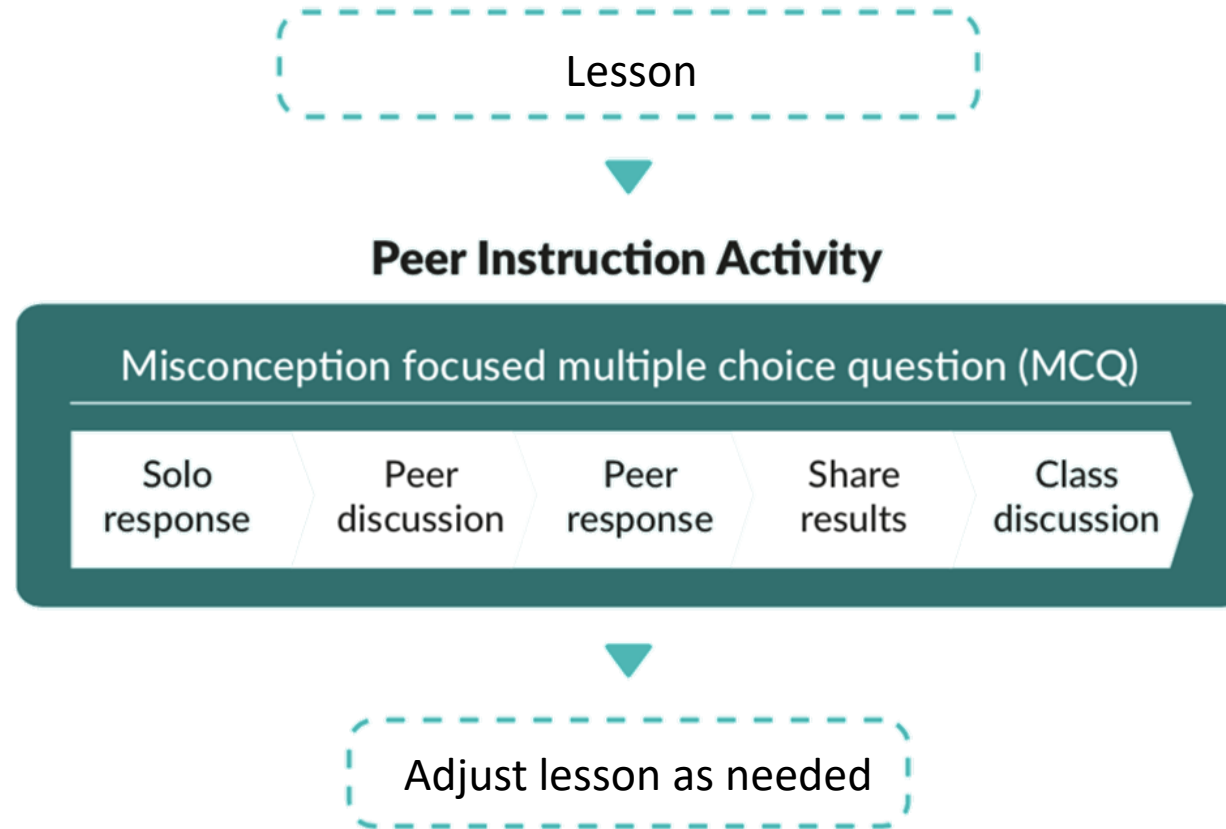


Assessment FOR Learning: Key influencing factors

1. Providing **effective feedback** to students.
2. Students' **active involvement** in their own learning.
3. **Adjusting teaching** to take into account the results of assessment.
4. Recognizing the profound **influence of assessment on students' motivation and self-esteem** - both crucial influences on learning.
5. Ensuring **students assess themselves and understand how to improve.**

**What is an example of
assessment FOR learning?**

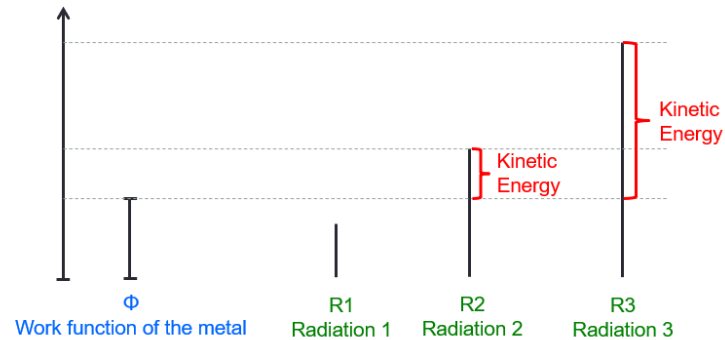
Assessment FOR Learning Strategy: Peer Instruction



Think about it!

- Which radiation will lead to the fastest electron?

Energy



Lesson on photoelectric effect

Peer Instruction Activity

Misconception focused multiple choice question (MCQ)

Solo
response

Peer
discussion

Peer
response

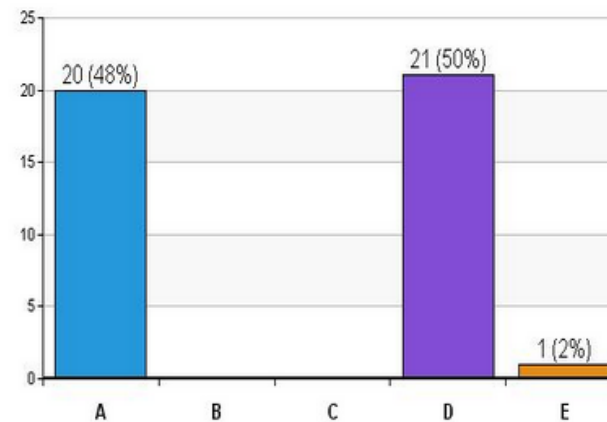
Share
results

Class
discussion

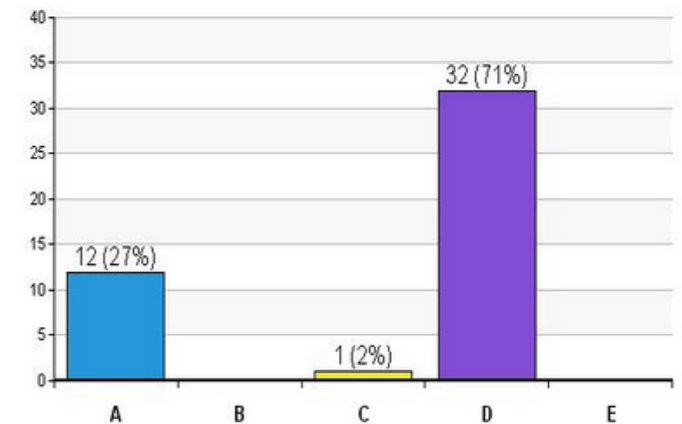
Clicker Question

- Which of the following metal has the highest work function?

A. Sodium	532nm
B. Zinc	290 nm
C. Copper	279 nm
D. Platinum	199 nm
E. Calcium	441 nm

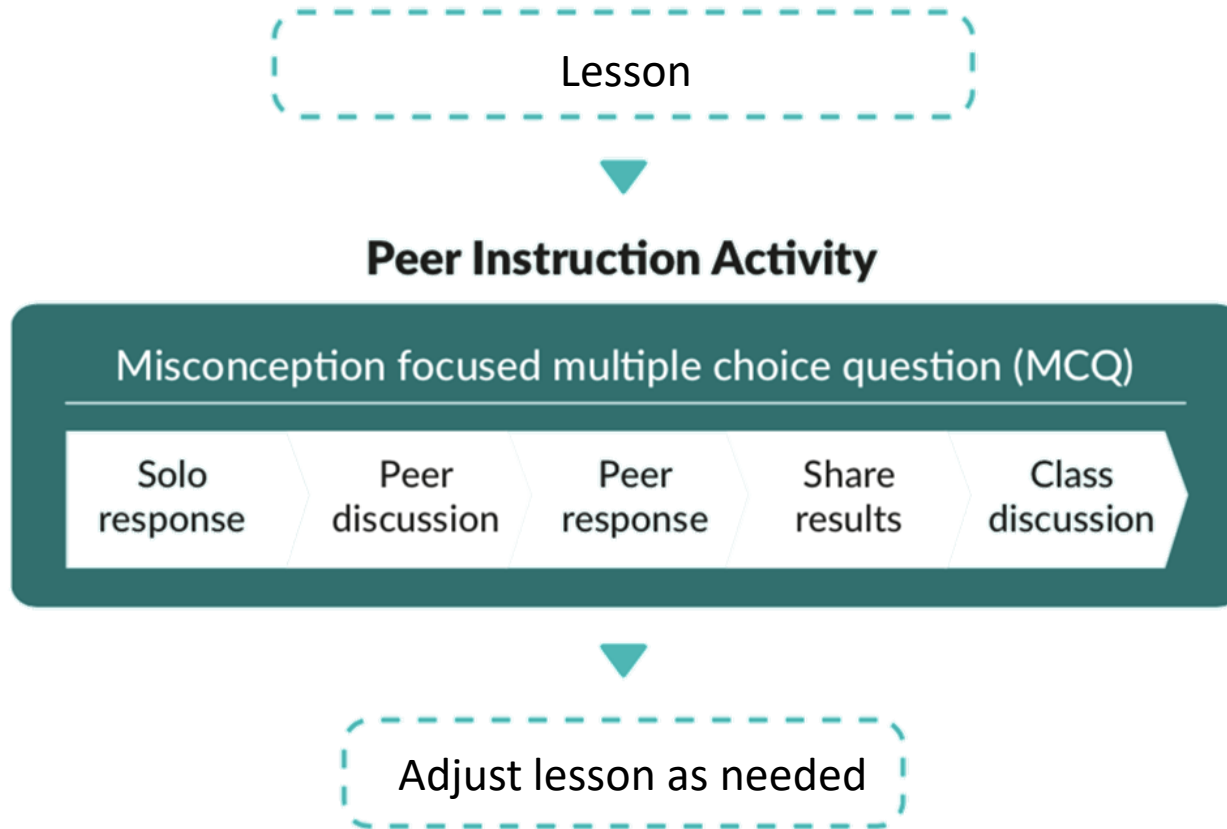


Solo responses



Peer responses

Assessment FOR Learning Strategy: Peer Instruction



1. Providing **feedback** to students. ✓
2. Students' **active involvement** ✓
3. **Adjusting teaching** ✓
4. Recognizing the **influence of assessment on students' motivation and self-esteem** ✓
5. Ensuring **students assess themselves and understand how to improve.** ✓

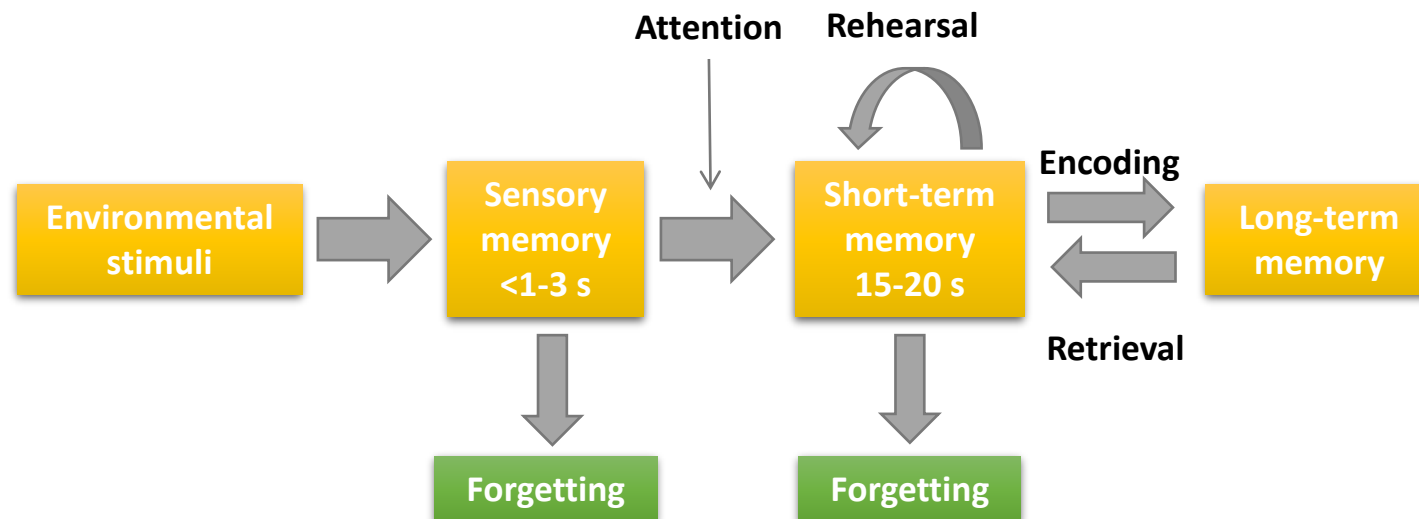
What are the evidence supporting Peer Instruction?

Peer Instruction is supported by extensive empirical evidence

1. Peer Instruction is effective in **promoting students' conceptual understanding** in a variety of STEM disciplines and courses across various institutions.
2. Peer Instruction can enhance both **qualitative and quantitative problem-solving skills**.
3. Peer Instruction **increases retention and lowers failure rates**.
4. Students have neutral to positive views on PI and seem to recognize its value over traditional teaching.

Peer Instruction aligns with principles on how people learn

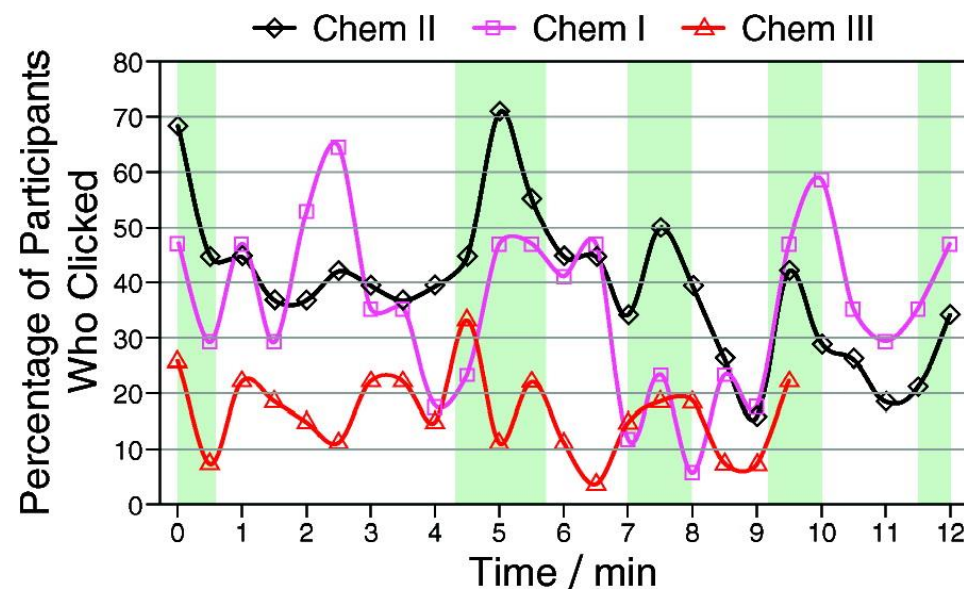
The human brain can only process small chunk of information at a time.



Peer Instruction aligns with principles on how people learn

The human brain can only process small chunk of information at a time.

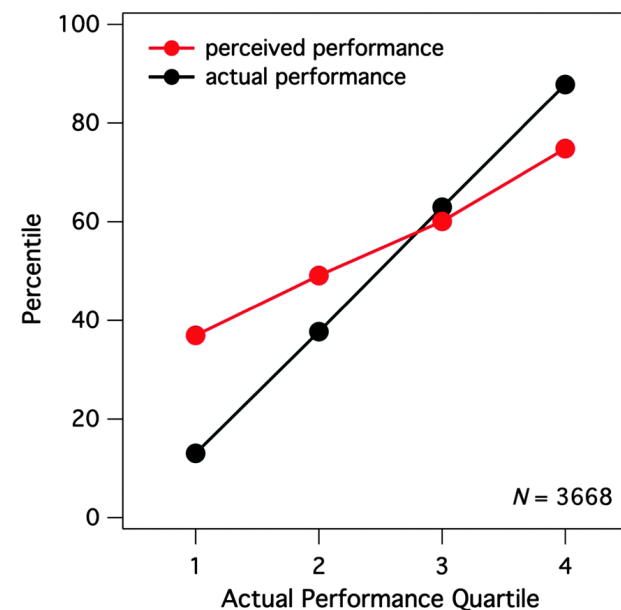
Within a 12 min lecture segment, students pay attention intermittently for about 7.5 min.



Peer Instruction aligns with principles on how people learn

It is challenging for human to accurately evaluate their competence.

“Low-performing students tend to overestimate their own performance while high-performing students tend to underestimate their performance.”



Peer Instruction aligns with principles on how people learn

Individuals are likely to learn more when they learn with others than when they learn alone.



Meaningful learning is facilitated by articulating explanations.

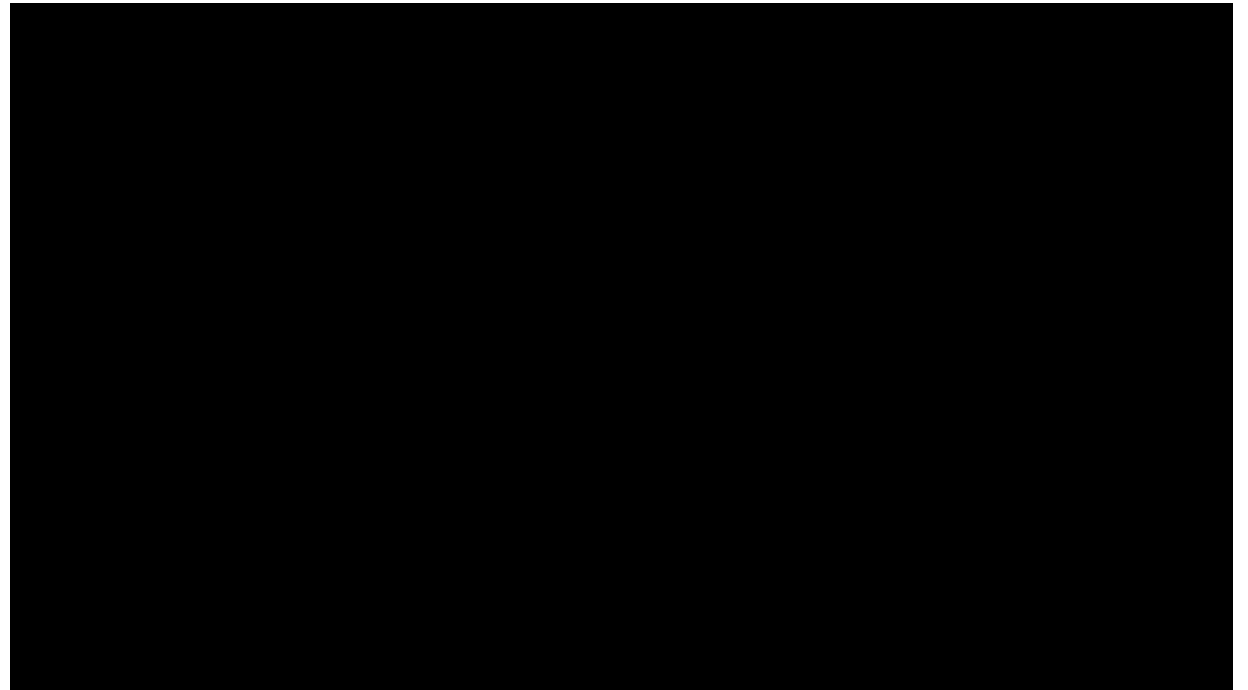
How to effectively implement Peer Instruction?

Peer Instruction versus clicker question

Clicker question

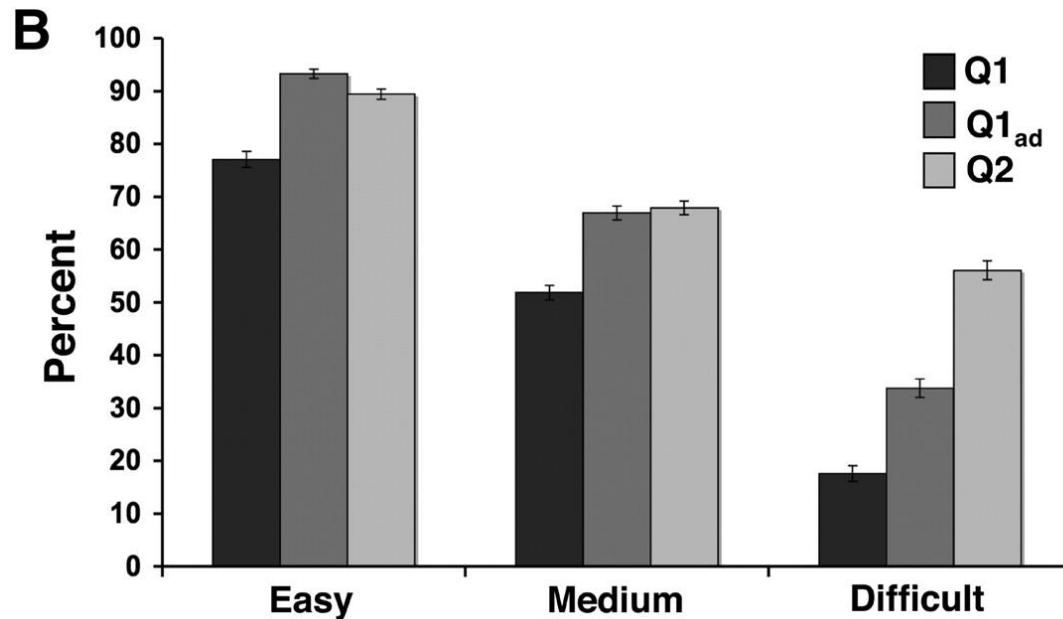


Peer Instruction



Peer Instruction is NOT a clicker question!

Questions should be challenging enough to provoke interest and discussion. More difficult questions promote greatest growth.



Q1: One question was voted on individually

Q1_{ad}: the same question was voted on again after peer discussion

Q2: A second isomorphic question addressing the same concept as the first question was voted on individually

Recall

The volume of a gas is inversely proportional to the pressure of a gas is known as

- A. Avogadro's Law
- B. Ideal Gas Law
- C. Charles's Law
- D. Boyle's Law
- E. Dalton's Law

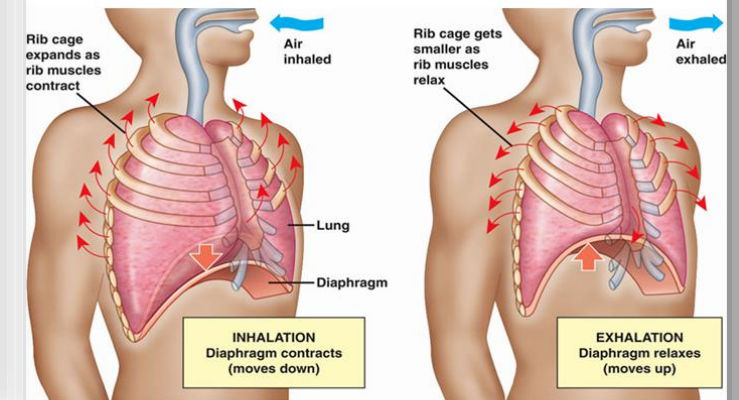
Algorithmic

A chamber has a pressure of 1.5 atm at a volume of 1.4 L. What is the pressure when the piston compresses the sample to a volume of 0.60 L?

- A. 3.5 L
- B. 0.64 L
- C. 1.8 L
- D. 1.6 L
- E. 0.29 L

Conceptual

The thoracic cavity expands during inhalation. In comparison to the atmosphere, would you expect lung pressure to

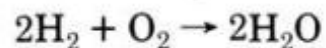


- A. increase
- B. decrease
- C. stay the same

Algorithmic competence \neq Conceptual understanding

Traditional Stoichiometry

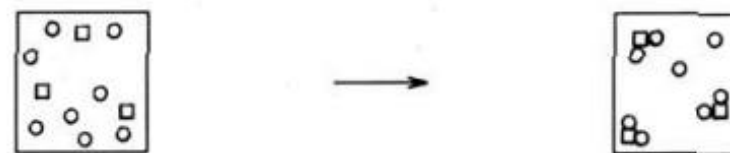
For a mixture of 2 mol H_2 and 2 mol O_2 reacting according to the following equation, what is the limiting reagent, and how many moles of the excess reactant would remain after the reaction is completed?



	Limiting Reagent	Excess Reactant Remaining
(a)	O_2	1 mol O_2
(b)	O_2	1 mol H_2
(c)	H_2	1 mol O_2
(d)	H_2	1 mol H_2
(e)	No reaction occurs since the equation does not balance with 2 mol H_2 and 2 mol O_2 .	

Conceptual Stoichiometry

The reaction of element X (\square) with element Y (\circ) is represented in the following diagram. Which of the equations best describes this reaction?



- (a) $3\text{X} + 8\text{Y} \rightarrow \text{X}_3\text{Y}_8$
- (b) $3\text{X} + 6\text{Y} \rightarrow \text{X}_3\text{Y}_6$
- (c) $\text{X} + 2\text{Y} \rightarrow \text{XY}_2$
- (d) $3\text{X} + 8\text{Y} \rightarrow 3\text{XY}_2 + 2\text{Y}$
- (e) $\text{X} + 4\text{Y} \rightarrow \text{XY}_2$

Algorithmic competence \neq Conceptual understanding

Table 1. Rate of Student Success

	<i>N</i>	Conceptual Question	Traditional Question	Level of Significance
<i>Gas Laws</i>				
All Students	285	31.23 %	87.72 %	$\ll 0.001$
Upper 27 % of class ^a	77	44.16 %	96.10 %	$\ll 0.001$
Lower 27 % of class ^a	77	14.29 %	80.52 %	$\ll 0.001$
<i>Stoichiometry</i>				
All Students	323	11.46 %	66.25 %	$\ll 0.001$
Upper 27 % of class	87	20.69 %	90.80 %	$\ll 0.001$
Lower 27 % of class	87	3.45 %	45.98 %	$\ll 0.001$

^a The upper and lower 27 % of the class refer to overall performance on the entire 25-item multiple-choice exam.

Probing for conceptual understanding

Venn Diagrams: Students compare, distinguish between

Graphical: Students estimate, examine, interpret

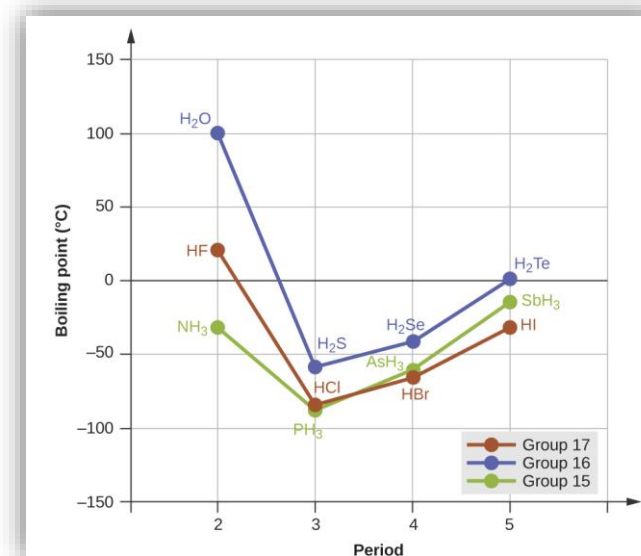
Pictorial: Interpret, explain, determine relationships

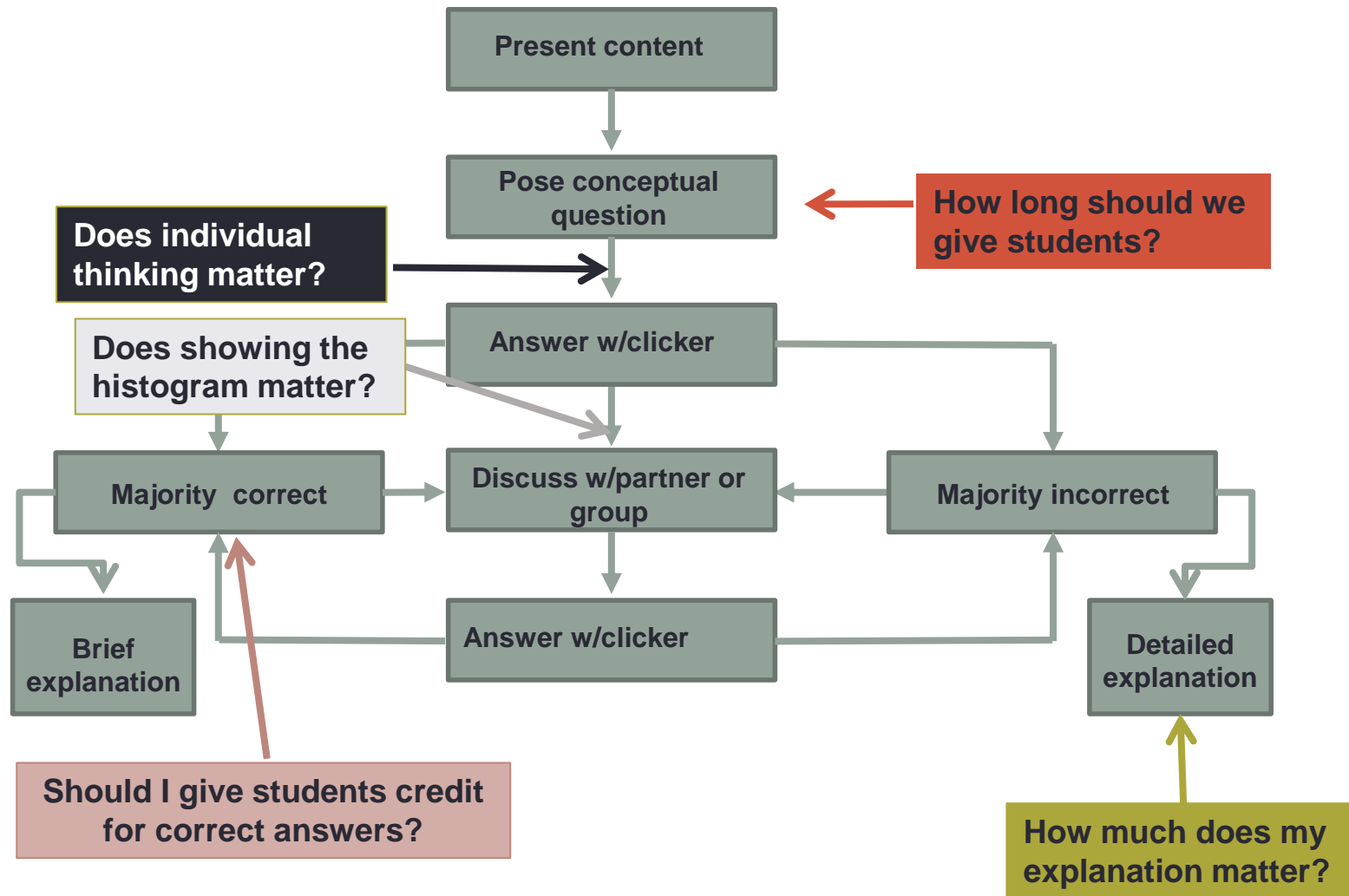
Perturbation of a system: Predict, relate, assess

Simulations



Others – categorization, distinguishing, prioritizing/ranking,





Voting methods



Learning Catalytics

What are the typical pitfalls with Peer Instruction?

Ensure alignment between your assessments classroom activities and learning goals!



Students' resistance

Explain why you are using this strategy on the syllabus, on day one, and several more times throughout the first third or so of the semester.



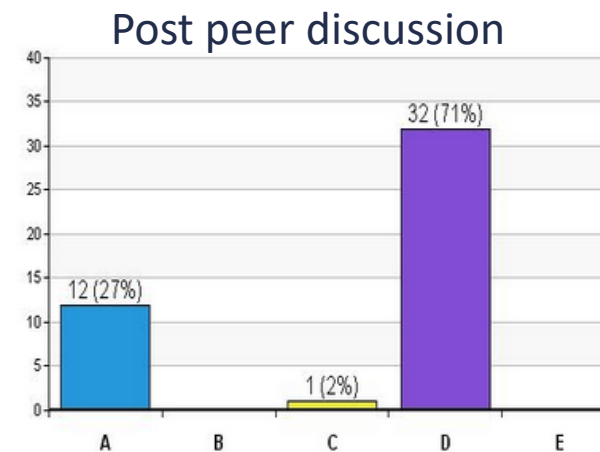
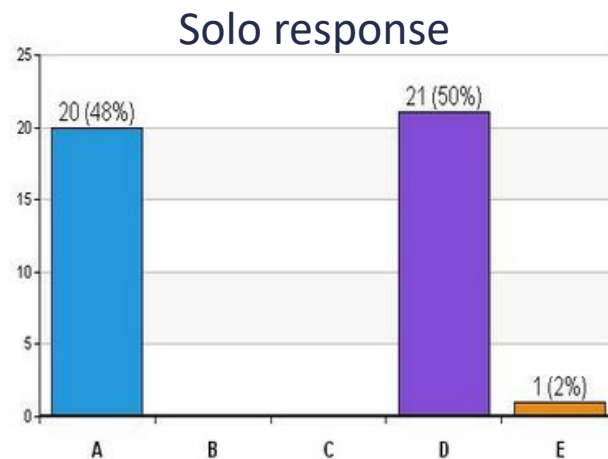
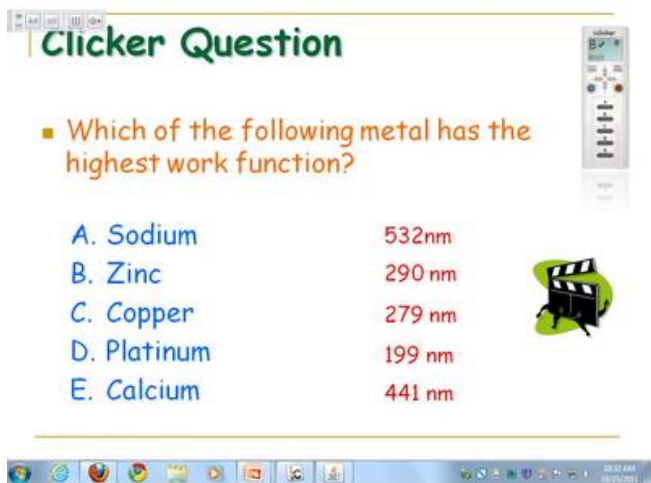
Students' resistance

Emphasize how much they learn when working together compare to when just listening to you.

Clicker Question

■ Which of the following metal has the highest work function?

A. Sodium	532nm
B. Zinc	290 nm
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E. Calcium	441 nm



Issues with fidelity of implementation

No one size fits all but adaptations should be purposeful and tested.

	Peer Instruction (N=195)
I used it basically as described by the developer.	16.9%
I made some relatively minor modifications	35.9%
I used some of the ideas, but made significant modifications	41.0%
I am not familiar enough with the developer's description to answer this question	6.2%
All users	100%



Henderson, C., & Dancy, M. H. (2009). Impact of physics education research on the teaching of introductory quantitative physics in the United States. *Physical Review Special Topics-Physics Education Research*, 5(2), 020107.

Andrews, T. M., Leonard, M. J., Colgrove, C. A., & Kalinowski, S. T. (2011). Active learning not associated with student learning in a random sample of college biology courses. *CBE—Life Sciences Education*, 10(4), 394-405.

Monitoring during peer discussion is essential!

- Enhance your **understanding of student thinking**
- Enhance **student-instructor relationship** especially in large classes
- It makes teaching a lot **more fun!**



Greg Nathan | University Communication | UNL

Gehrtz, J., Brantner, M., & Andrews, T. C. (2022). How are undergraduate STEM instructors leveraging student thinking?. *International Journal of STEM Education*, 9(1), 1-20.

<https://www.auburn.edu/cosam/aulap/>

<https://news.unl.edu/newsrooms/today/article/biology-education-research-earns-couch-national-award/>

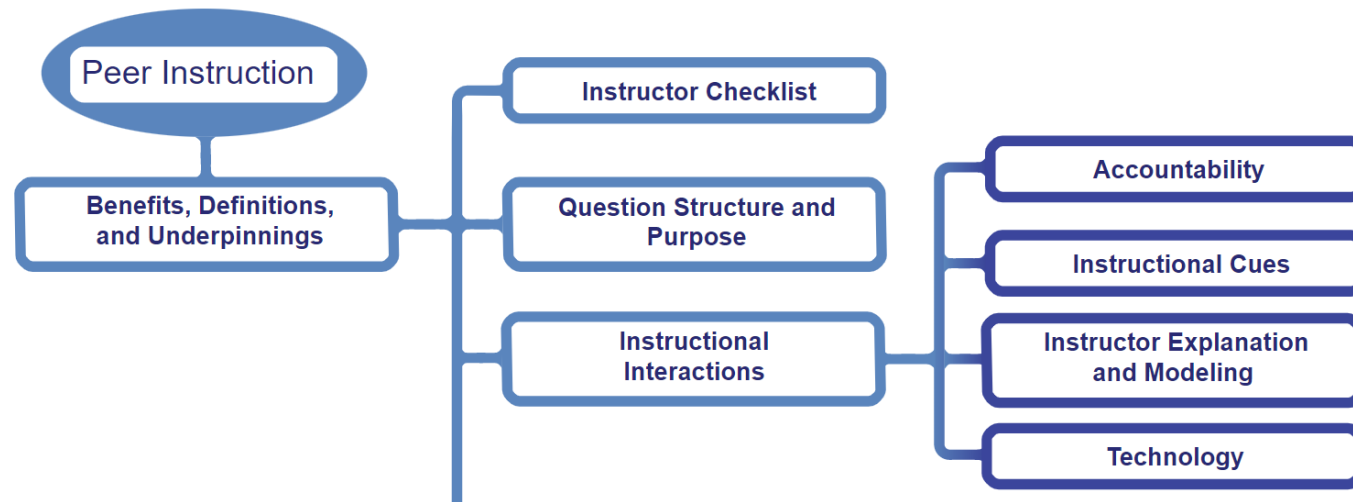
**What resources exist to help
me implement
Peer Instruction?**

Details on Peer Instruction



Evidence-Based Teaching Guides

Peer Instruction



<https://lse.ascb.org/evidence-based-teaching-guides/peer-instruction/>

Questions for Peer Instruction

- Aktiv: www.aktiv.com



- Vicente Talanquer – Chemical Thinking Interactives:
<https://sites.google.com/site/ctinteractives/>



- Stacey Lowery Bretz' concept inventories:
<https://sites.google.com/miamioh.edu/bretzsl/research/assessment-tools?authuser=0>



- PhET simulations – translated in numerous languages: <https://phet.colorado.edu/>



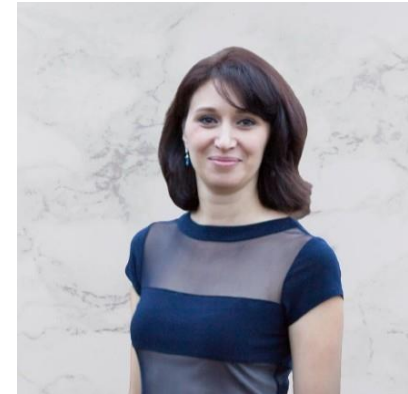
Thank you!



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